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SYNTHESIS AND PROPERTIES OF NEW DERIVATIVES OF MALEOPIMARIC AND CITRACONOPIMARIC ACIDS

The review article summarizes data on synthesis of new derivatives of maleopimaric acid (the most available diterpenoid acid, isolated from maleated rosin) – esters, amides, imides, imido amides, imido esters, diimido acids etc. and of oxygen- and nitrogen containing derivatives of previously unknown citraconopimaric acid. The article mainly covers the literature data since 2000 year as well as work carried out in the Institute of Chemistry of New Materials of NAS of Belarus. *Keywords:* rosin, terpenoids, maleopimaric acid, citraconopimaric acid, esters, amides, imides.

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СИНТЕЗ И СВОЙСТВА НОВЫХ ПРОИЗВОДНЫХ МАЛЕОПИМАРОВОЙ И ЦИТРАКОНОПИМАРОВОЙ КИСЛОТ

Представлен обобщающий и систематизирующий обзор данных по синтезу производных малеопимаровой кислоты (наиболее доступной дитерпеноидной кислоты, выделяемой из малеинизированной канифоли) – эфиров, амидов, имидов, имидоамидов, имидоэфиров, диимидокислот и др., а также кислород- и азотсодержащих производных ранее неизвестной цитраконопимаровой кислоты. Приведены литературные данные с 2000 г., а также работы, выполненные в Институте химии новых материалов НАН Беларуси.

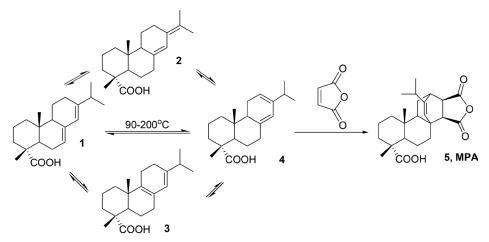
Ключевые слова: канифоль, терпеноиды, малеопимаровая кислота, цитраконопимаровая кислота, сложные эфиры, амиды, имиды.

Terpenes from wood resins (essential oils and rosins) constitute one of the valuable part of plant resources and main source of terpene substances. In the last decades, the intensive studies were directed toward the development of new methods for valuable chemical products preparation from renewable raw materials and the individual substances that can be isolated or synthesized from rosin are of special interest. The most promising approach to the preparation of individual diterpene derivatives is to modify rosin in particular by active dienophiles. The most available individual substance thus obtained is maleopimaric acid (MPA), isolated from rosin-maleic anhydride adduct (RMA). This compound is used for the production of printing inks, alkyd resins, lubricants, paper sizing agent, and its derivatives show pronounced biological activity [1–5]. This review article mainly covers the literature data on preparation of new maleo- and citraconopimaric acid derivatives since 2000 year as well as work carried out in the Institute of Chemistry of New Materials NAS of Belarus.

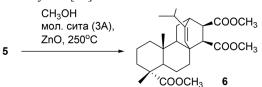
1. The synthesis and application of maleopimaric acid and its derivatives. The MPA, **5** was firstly obtained independently by Arbuzov and Ruzicka et al. in 1932 in the course of studies of resin acids structure [6, 7]. Two main groups of methods have been proposed for the MPA synthesis. The first group consists of methods using reaction of levopimaric acid **4** (content up to 53% in wood resin [8]) and maleic anhydride at 30–80 °C [9, 10]. The disadvantage of these methods is that abietic-type acids (abietic **1**, neoabietic **2**, palustric **3**) except levopimaric acid **4** stay intact decreasing MPA yield.

The second group of methods is based on reaction of abietic-type acids (abietic 1, neoabietic 2, palustric 3) of rosin with maleic anhydride at 90–200 °C or MW-irradiation including the application of catalysts followed by isolation of MPA from mixture [11–13]. These methods are the most often used for individual MPA synthesis.

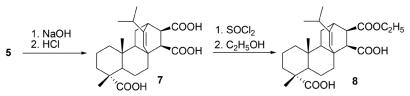
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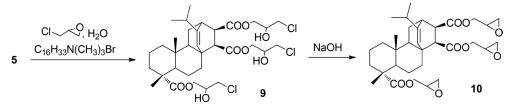
MPA is a tribasic acid anhydride, as a result its derivatives on carboxyl groups, in particular mono-, di-, tri- and polyesters are among the most important. MPA methyl ester was firstly obtained by treatment of MPA with diazomethane [14] and monoethyl-, propyl- and butyl esters were prepared by reaction of MPA with PCl₃ followed by treatment with the correspondent alcohol [15]. The MPA vinyl ester was synthesized by reaction of MPA and vinyl acetate catalyzed by mercury salt [16]. MPA trimethylester **6** was prepared by reaction of MPA and methyl alcohol in autoclave at the presence of molecular sieves (3Å) at 250 °C for 4 h in 98% yield [17].



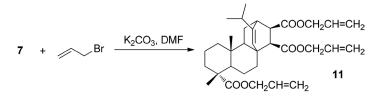
The treatment of MPA with 30% NaOH solution and subsequent acidification gave triacid 7. Reaction of the acid 7 with $SOCl_2$ followed by the treatment with ethyl alcohol led to the regioselective formation of monoethyl ester 8 in 73% yield [18].



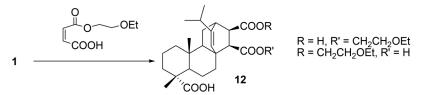
The MPA triglycidyl ester useful for polymer producing was prepared in two stages. Firstly the reaction of MPA with epichlorohydrin in the presence of hexadecyltrimethylammonium bromide gave triol **9**, which was converted into triglycidyl ester **10** in 87% yield under the treatment with sodium hydroxide solution [19].



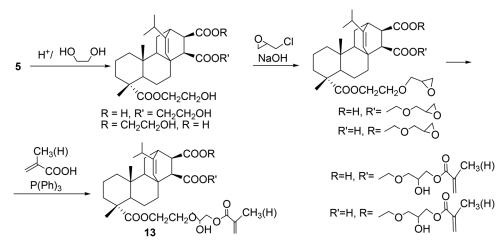
MPA triallyl ester **11** was synthesized by reaction of triacid **7** with allyl bromide in 91% yield and used for preparation of bio-based thermosetting resins [20].



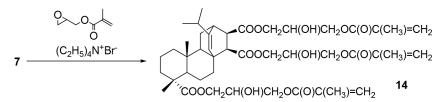
The mixture containing the isomeric MPA 2-ethoxyethyl esters **12** was prepared by reaction of tall rosin with 2-ethoxyethylmaleat and used as paper sizing agent [21].



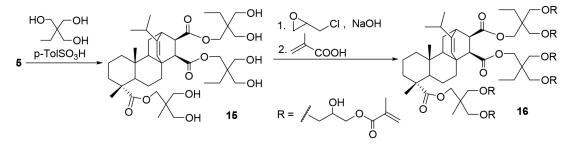
MPA was used for the synthesis of acrylic esters. The esterification of MPA with ethylene glycol gave a mixture of diesters that on treatment with epichlorohydrin followed by reaction with acrylic or methacrylic acids in the presence of $P(Ph)_3$ provided the mixture of acrylic esters **13** useful for corrosion protection of metals [22].



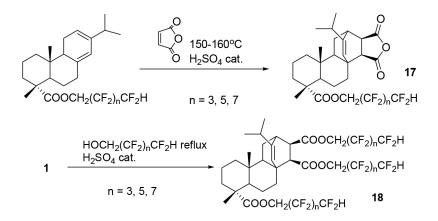
The MPA triacrylic ester 14 was prepared in two stages. MPA was converted into triacid 7, and subsequent treatment of the latter one with glycidyl methacrylate in the presence of tetraethylammonium bromide gave triester 14 useful for producing of methyl methacrylate and stirol copolymers [23].



RMA was esterified by trimethylol propane to give mixture containing polyol **15**. Reaction of polyol **15** successively with epichlorohydrin and methacrylic acid afforded polyacrilic ester **16**. Thermal, mechanical and chemical performance of vinyl ester resins based on ester **16** as high performance coatings materials were found to be comparable to those of conventional petroleum based coating materials [24].

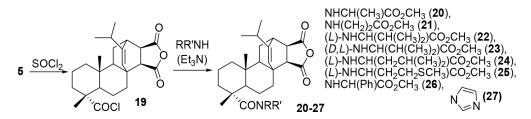


The perfluoroalkyl esters of MPA **17** were prepared by Diels-Alder reaction of rosin perfluoroalkyl esters with maleic anhydride at 150–160 °C under acidic catalysis [25]. The refluxing of MPA in the corresponding alcohol at the presence of sulfuric acid gave MPA perfluoroalkyl triesters **18** in 55–70% yield [26].

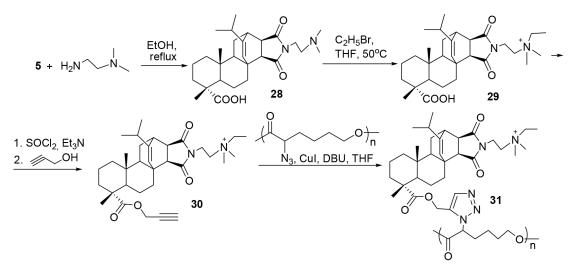


Amide compounds from MPA were firstly synthesized by treatment of MPA vinyl ester with cyclohexylamine [16]. It was found substituted MPA amides show pronounced hepatoprotective [27], antimicrobial [5] and fungicidal properties [28], and MPA based polyamides are suitable for applications as protective films, printing inks, thermo stable polymers [29–31].

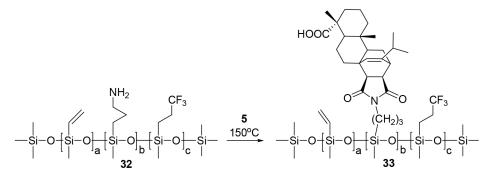
MPA amides **20–27** with anti-inflammatory and antiulcer properties were synthesized by reaction of MPA chloride **19** with amino acids and imidazole. It was established that amide **24** prepared from L-leucine has the highest anti-inflammatory properties and show low toxicity [32].



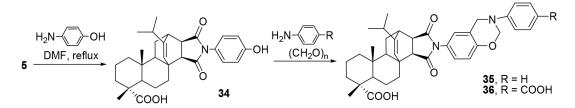
Wang et al. synthesized antimicrobial polymers **31** based on MPA N-(2-N,N-dimethylethyl)imide and azide substituted poly(ε -caprolactone). MPA N-(2-N,N-dimethylethyl)imide after quaternization and esterification with propargyl alcohol was grafted on polymer by click-reaction with the use of CuI/ DBU as catalysts [33].



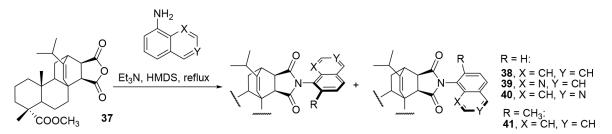
The MPA imide modified fluorosilicone resin **33** was prepared by heating of MPA and amine containing fluorosilicone resin **32** at 150°C for 4 h. This novel crosslinker was added into high temperature vulcanization rubber composition to prepare a series of MPA modified fluorosilicone rubbers with increased mechanical properties [34, 35].



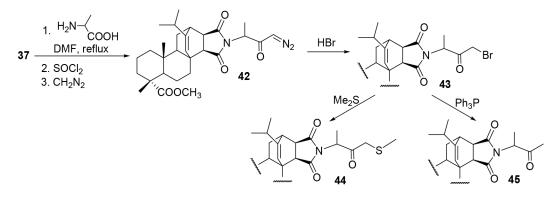
The benzoxazine monomers **35**, **36** were synthesized from MPA in two steps. The reaction of MPA and 4-aminophenol afforded MPA N-(4-hydroxy)phenyl imide **34** that on reaction with paraformalde-hyde and aniline (or 4-aminobenzoic acid) gave benzoxazines **35**, **36**. The polybenzoxazines were prepared by polymerization of compound **35**, **36** and showed high thermal stability [36].



Reactions of MPA methyl ester **37** with 1-naphtylamine, 8-aminoquinoline, 5-aminoisoquinoline [37] and 2-methy-1-naphtylamine [38] in refluxing triethylamine in the presence of hexamethyldisilazane (HMDS) give MPA imides **38–41** as the mixture of atropisomers. It was found N-(2-methylnaphtyl-1) imide of MPA methyl ester **41** has anti-proliferative properties and a significant difference in the level of cytotoxicity was observed between R- and S-conformers [38].

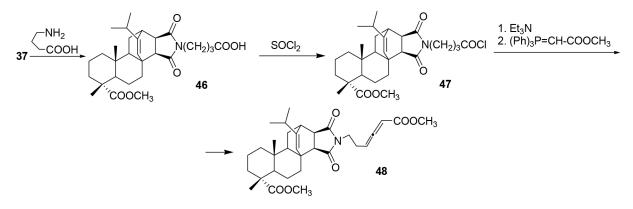


The bromomethylketone 43 was synthesized for the first time under Arndt–Eistert reaction conditions via condensation of ester 37 and alanine. Treatment of ketone 43 with Me_2S or Ph_3P led to the formation of methylsulfide 44 or to reduction into ketone 45, presumably due to the traces of water [39].

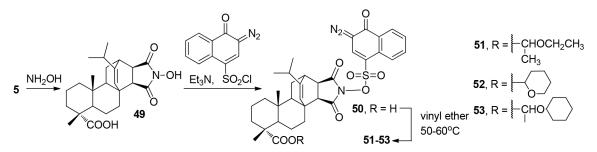


The allene **48** was synthesized from methyl maleopimarate **37** and γ -aminobutyric acid (GABA). Direct fusion or condensation of ester **37** with GABA gave the imide **46** in 52 or 72% yields, respectively, which was converted into the corresponding acid chloride **47** by reaction with SOCl₂ excess.

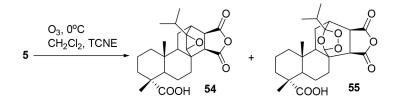
Treatment of chloride with Et_3N and subsequent reaction with methyl(triphenylphosphoranylidene)acetate provided the corresponding allenoate **48** in 66% yield [40].



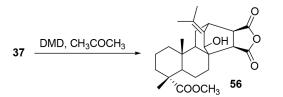
New molecular glass compounds containing 2-diazo-1-naphthoquinone-4-sulfonyl group were designed based on MPA. Reaction of MPA with hydroxylamine gave MPA N-hydroxyimide **49** in 70% yield that on esterification with 2-diazo-1-naphthoquinone-4-sulfonyl chloride provided N-hydroxy maleopimarimide sulfonate **50**. The carboxylic acid group of the acid **50** was then protected by the reaction with vinyl ether compounds (vinylethyl ether, cyclohexylvinyl ether, dihydropiran) to give the corresponding esters **51–53**. The novel one-component molecular glass resists **51–53** have potential to become high-performance i-line photoresists [41].



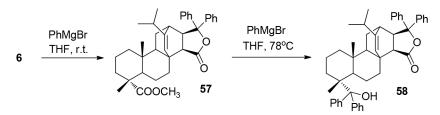
Hess et al. showed that the ozonolysis of MPA in the presence of tetracyanoethylene (TCNE) and subsequent purification lead to isolation of known epoxide **54** in 20% yield and the ozonide **55** in 7% yield [42].



Reaction of the MPA methyl ester **37** and dimethyldioxirane (DMD) resulted in regioselective oxidation of the bridging double bond to form the alcohol **56** [43].

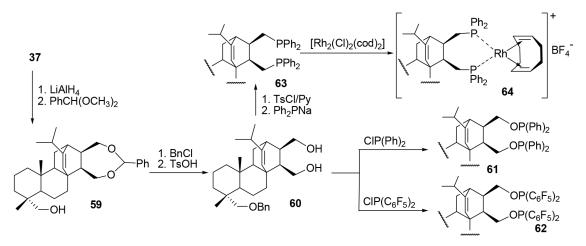


Yao et al. established that reaction of MPA trimethyl ester **6** with PhMgBr proceeds at only one carboxyl group at r. t. to form lactone **57** that on treatment with Grignard reagent at 78°C converts into alcohol **58** [44].

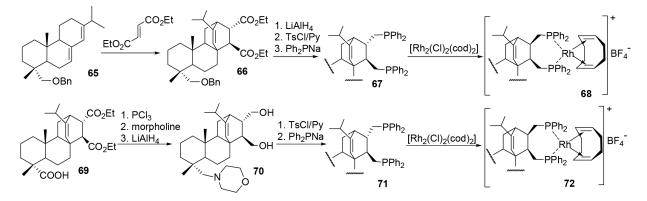


Asymmetric synthesis has been one of the most important topics in organic chemistry over the past decades and terpenoid substances prepared from rosin are very accessible optical pure compounds that make them useful starting synthone for synthesis of chiral ligands.

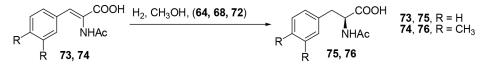
Chiral bisphosphinic ligand **63**, **67**, **71** were synthesized from MPA [45]. Reduction of MPA methyl ester **37** with LiAlH₄ and protection of 1,4-diol moiety gave benzylidenedihydroxy derivative **59**. Formation of benzylic ether and deprotection led to the diol **60** that was treated with phosphine chlorides to give 1,4-bisphosphineoxides **61**, **62**. Treatment of diol with *p*-toluenesulfochloride gave ditosylate that on reaction with sodium diphenylphosphide afforded bisphosphine **63**. Reaction of bisphosphine with di-**µ**-chloro-bis-(1,5-cyclodiene)dirhodium and NaBF₄ in CH₂Cl₂ yields the cation Rh(I)complexes **64**.



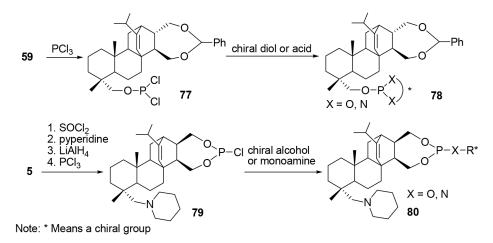
The ligands **67**, **71** and their Rh(I)-complexes **68**, **72** were prepared from benzylic ether of abiethynol **66** and diethyl fumaropimarate **69** in a similar manner [45].



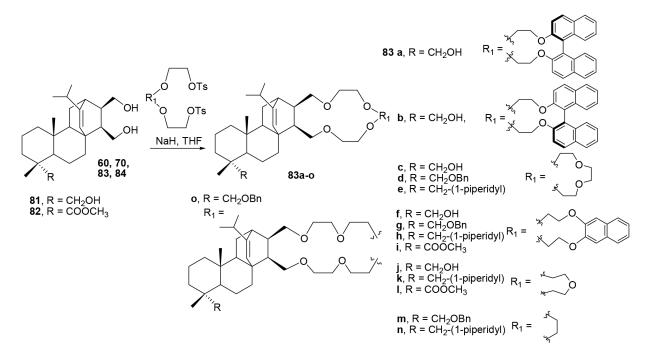
The rhodium complexes **64**, **68**, **72** were used as catalysts for hydrogenation of substituted cinnamic acids **73**, **74**. It was found application of these catalysts provided hydrogenation to give amino acids **75**, **76** with up to 100% yield and 66% *ee* [45].



Two chiral phosphorus derivatizing agents 77 and 79 were synthesized from MPA and it was established these derivatives are well suited for the ³¹P NMR-based determination of enantiomeric excess of chiral alcohols and amines in solutions [46].



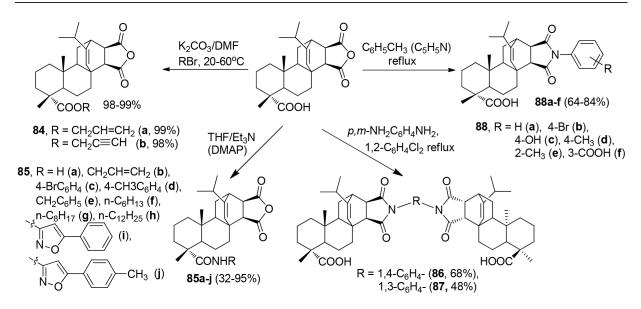
A series of chiral crown ethers **83a–o** including 28-crown-8, 22-crown-6, 20-crown-6, 17-crown-5 and 14-crown-3 bearing hydroxyl side group derivatives comprising of binaphthyl, naphthalene and ros-in acid moieties in the crown ring was prepared based on MPA [47, 48].



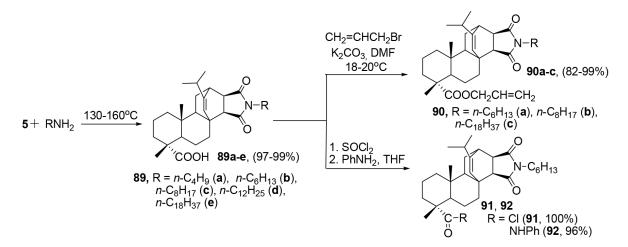
These receptors **83a–o** showed strong affinity and different complementarity for various amine salts, and exhibited excellent enantiodiscriminating abilities towards protonated primary amines and amino acid methyl ester salt isomers in chiral recognition [48].

To conclude this section it is worth noting that apart from individual MPA derivatives the esters of RMA with polyols (ethylene glycol, glycerol, pentaerythritol etc.) as well as RMA imides are widely used in industry for production of alkyd and epoxy resins, printing inks, polymers and copolymers, sizing agents for paper making [3, 4].

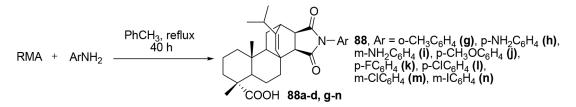
2. Researches on diterpenoids at The Institute of Chemistry of New Materials of the National Academy of Sciences of Belarus. The new preparative methods for synthesis of MPA unsaturated esters 84a, b, amides 85a–j, imides 88a–f and diimido diacids 86, 87 were developed by reaction of MPA with allyl-, propargyl bromide, aliphatic and aromatic amines [49–52].



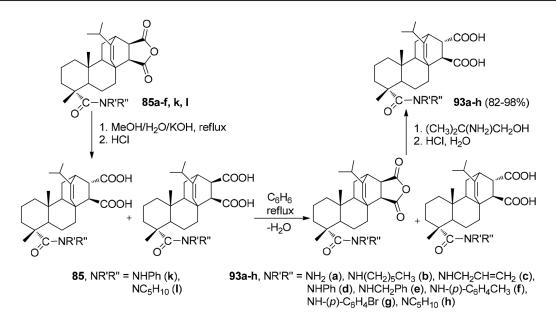
The synthesis of previously unknown MPA N-alkylimides **89a–e**, including long-chain dodecyl- and octadecylimides **89d**, **e** by reaction of primary aliphatic amines with MPA in the melt was developed. The method eliminates the use of organic solvents, allows application as a starting compound MPA and acetic acid solvate, significantly reduces reaction time [53]. Imido amide **92** was prepared by reaction of imide **89b** with thionyl chloride followed by reaction with aniline.



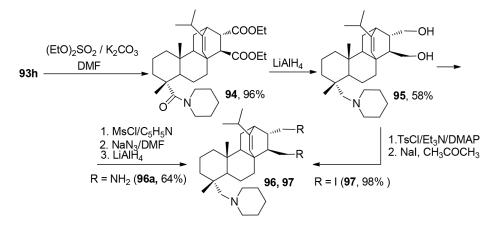
The MPA N-arylimides synthesis directly from RMA without isolation of individual MPA was developed based on treatment of RMA with primary aromatic amines in toluene (refluxing) to form exclusively N-arylimides **88a–e**, **g–n** in preparative yields (60–99% based on MPA contained in the RMA) [51].



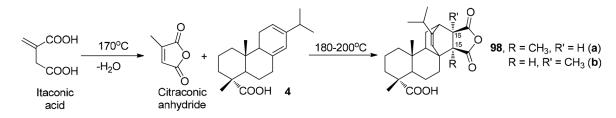
We elaborated the method of fumaropimaric acid monoamides **93a–h** synthesis containing *trans*-1,2-dicarboxylic fragment by alkali hydrolysis of anhydride group in MPA amides **85a–f**, **k**, **l** and subsequent isomerization of *cis*-1,2-dicarboxylic acid. The monoamides **93a–h** were isolated from the reaction mixture by conversion into water-soluble ammonium salts [54].



Esterification and subsequent reduction of amide **93h** afforded diol **95** that was converted into diamine **96** and diiodide **97** by conventional methods.

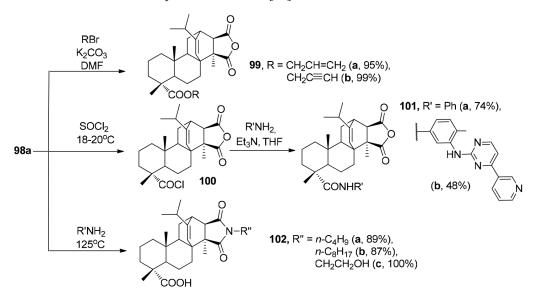


With a goal of citraconopimaric acid (CPA) synthesis the reaction of citraconic anhydride, formed *in situ* from readily available itaconic acid (produced industrially by the fermentation of carbohydrates) on heating above the melting point (172 °C), and rosin has been studied. It was established that the heating of mixture of rosin and itaconic acid at 180–200 °C for 8–12 h leads to the reaction of citraconic anhydride with levopimaric acid and to the formation of complex mixture containing isomeric citraconopimaric acids **98a**, **b** and unreacted resin acids (IR, ¹H NMR data). IR-spectra of the mixture contain characteristic bands of anhydride cycle at 1850, 1785 cm⁻¹. The crystallization of the reaction product from CCl₄ at 18–20 °C gives precipitate of solvate CPA with CCl₄, and its thermal decomposition at 130 °C gives CPA as a mixture of two isomers **98a**, **b** at ratio ~1:0.36. CPA isomers differ in the arrangement of the methyl group in the anhydride cycle only and have almost the same chromatographic mobility in different systems. By partial crystallization of mixture of isomers **98a**, **b** [55].



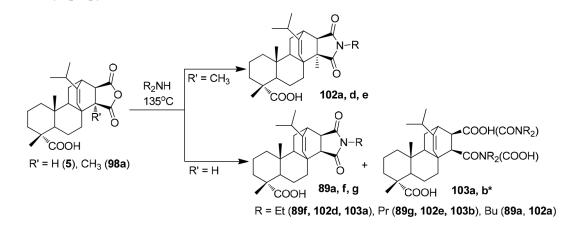
The full assignments of peaks in ¹H and ¹³C NMR spectra were performed by two-dimensional NMR spectroscopy, and it was confirmed that the compound **98a** is the adduct of citraconic anhydride and levopimaric acid (citraconopimaric acid) and a close structural analog of MPA, containing methyl in the α -position of the anhydride group [56].

Reaction of CPA **98a** with allyl- or propargylbromide in the presence of K_2CO_3 afforded unsaturated esters **99a**, **b** [49]. The treatment of CPA **98a** with thionyl chloride gave the acid chloride **100**, and its reaction with aniline and N-(2-methyl-5-aminophenyl)-4-(pyrid-3-yl)pyrimidine-2-amine led to the formation of anilide **101a** and heterocyclic amide **101b** [57].



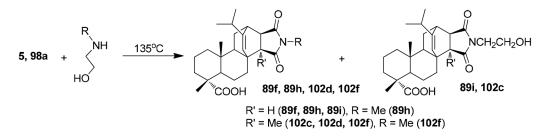
The N-alkylimides **102a–c** were synthesized by reaction of CPA with secondary *n*-butylamine, *n*-octylamine, ethanolamine at 125°C [58]. The treatment of CPA N-(2-hydroxyethyl)imide **102c** with dimethylsulfate in the presence of K_2CO_3 gave methyl ester of CPA N-(2-hydroxyethyl)imide (yield 73%).

Reactions of CPA and MPA with secondary aliphatic amines (diethyl-. dipropyl-, dibutylamine) were investigated for the first time [59]. Formation of CPA N-substituted imides instead of expected amido acids was found. In contrast to CPA reaction of MPA gives the only product in reaction with dibutyl-amine, but in reactions with diethyl- and dipropylamines amido acids **103a**, **b** were formed together with MPA N-ethyl(propyl)imide.

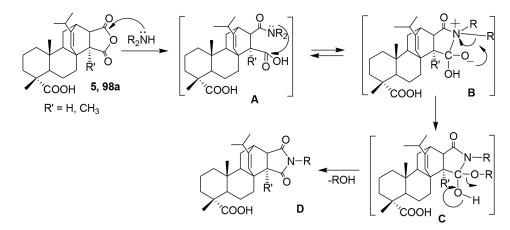


^{*} Exact location of the amide group in compounds 103 a, b is not defined.

The treatment of MPA and CPA with unsymmetrical methyl(ethyl)-2-hydroxyethylamines leads to formation of two imides: N-methyl(ethyl)- and N-(2-hydroxyethyl)imides of MPA and CPA (in molar ratio 1:(0.78–1.63)) [59].



The mechanism of this reaction presumably involves the formation of the intermediate amic acid (A) that undergoes intramolecular cyclization to form the cyclic intermediate (B), which is stabilized by the migration of the alkyl group (C), followed by elimination of alcohol and formation of the cyclic N-al-kylimid (D). In the case of the reaction MPA, CPA with unsymmetrical secondary amines the formation of two imides, presumably, the result of migration of various alkyl groups in the intermediate (B). The mechanism of this reaction is likely close to the mechanism of N-substituted imides formation in reaction of cyclic anhydrides with primary amines.



Ways of practical use of MPA and CPA derivatives. The effect of the MPA allyl ester 84a was studied on rheological properties in modifying of low density polyethylene (LDPE) with itaconic acid in the presence of free radical initiators. It was found that ester 84a additive in an amount of 1 wt. % of the reaction system decreases the efficiency of grafting only in 13% with more than 2 times the viscosity of the grafted product is reduced. The MPA ester 84a may be used as a viscosity regulator during reactive extrusion of functionalized polyethylene with the desired rheological performance as well as in the preparation of coextruded multilayer polymeric products. Adhesive was developed based on ethylene-vinyl acetate copolymer (CEVA) with the addition of the MPA allyl ester 84a, including CEVA, fumaric acid, an organic peroxide, a stabilizer and 0.01–1.0% of allyl ester 84a. Adhesive bond strength with polar substrate of polyamide-6 for developed adhesive reaches 4.9 kN/m, which is 91% higher than that for similar adhesive compounds [60].

The MPA N-hexylimide **89b** is the effective adhesive additive to LDPE in the preparation of composite materials to wood at a concentration of 3%. The application of imide **89b** retains strength, highly elastic and rheological properties of the polymer composition, simultaneously the adhesive strength of the LDPE to the wood increases to 3.6 MPa, which is 1.6–1.8 times greater in comparison with samples of pure LDPE [61].

It was found that the nitrogen-containing MPA derivatives (aromatic amides **85a–c**, **f**, imide **89e** etc.) have twisting power and can be used as effective additives to nematic LC-materials [62].

A fungicidal composition comprising a salt of MPA N-hexylimide and 8-hydroxyquinoline was designed to protect the cellulose-containing materials [63].

Films of the compounds **5**, **84a**, **b**, **98a**, **99a**, **b** prepared by thermal vacuum deposition method (TVD) are sensitive to UV radiation (3.6–7.1 J/cm²) and of practical interest for the production of photomasks for wet etching topology in transparent conductive layers of indium tin oxide (ITO) [49].

To conclude this short review, since the last decades there has been a trend towards the development of methods for the use of renewable raw materials instead of fossil in the manufacture of industrial products. As it is evident from the review, it is possible to obtain a great number of based on rosin substances with diverse structures and properties for various applications, and these works constitute important part of the green chemistry development.

References

1. Galin F. Z., Flehter O. B., Tret'iakova E., "Synthesis and reactions of diene adducts of resin acids", *Himiia i komp'iuternoe modelirovanie. Butlerovskie soobshcheniia* [Chemistry and computer simulation. Butlerov communications], 2004, no. 2, pp. 1–21.

Wiyono B., Tachibana S., Tinambunan D., "Characteristics and chemical composition of maleopimaric and fumaropimaric rosins made of indonesian Pinus merkusii rosin", *Pakistan Journal of Biological Sciences*, 2007, vol. 10, no. 18, pp. 3057–3064.
Yoshi K., Ken H., Kyosuke B., Masahiko S., Arakawa Chem Ind., *Rosin emulsion sizing agent and paper*, CN, Pat.

№ 105386366, 2016.

4. Zhang J., Rosin-based Chemicals and Polymer, Smithers Rapra Technology, Akron, USA, 2012.

5. Svikle D. Ya., Prikule A. Ya., Shuster Ya. Ya., Veselov I. A., "Biological Activity and Toxicity of Maleopimaric Acid Derivatives", *Pharmaceutical Chemistry Journal*, 1978, vol. 12, no. 5, pp. 617–620.

6. Arbuzov B. A., "About the presence of conjugated double bounds in abietic acid", *Zhurnal obshchei khimii* [Journal of General Chemistry], 1932, vol. 2, no. 2, pp. 806–809.

7. Ruzicka L., Ankersmit P. J., Frank B., "Polyterpene und Polyterpenoide LXXIII. Anlagerung von Maleinsäureanhydrid an Abietinsäure und Dextropimarsäure", *Helvetica Chimica Acta*, 1932, vol. 15, no. 1, pp. 1289–1294.

8. Komschilov N. F., Kanifol', ee sostav i stroenie smolianykh kislot [Rosin, its composition and structures of resin acids], Lesnaia promyshlennost', Moscow, RU, 1965.

9. Sokolov A. G., *Sposob kompleksnoi pererabotki sosnovoi zhivitsy* [The way to complex treatment of pine resin], Komitet po delam izobretenii i otkrytii pri Sovete Ministrov SSSR, USSR, Pat. 113132, 1958.

10. Lloyd W. D., Hedrick G. W., "The Diels-Alder reaction of levopimaric acid and its use in quantitative determination", *Journal of Organic Chemistry*, 1961, vol. 26, no. 6, pp. 2029–2032.

11. Aldrich P. H., Process for separation of rosin adducts from mixtures with rosin, US, Pat. № 3562243, 1971.

12. Gonis G., Slezak F. B., Lawson N. E., "Preparation of maleopimaric acid", *Industrial and Engineering Chemistry Product Research and Development*, 1973, vol. 12, no. 4, pp. 326–327.

13. Wang, H. X., Shang, S. B., Li, J. F., "Synthesis of maleopimaric acid under microwave irradiation", *Chemical reagent*, 2009, vol. 33, no. 3, pp. 177–179.

14. Ruzieka L., LaLande W. A., "Zur Kenntnis der Diterpene. (44. Mitteilung). Über die Einwirkung von Ozon und Permanganat auf das Anlagerungsprodukt von Maleinsäure-anhydrid an Lävopimarsäure", *Helvetica Chimica Acta*, 1940, vol. 23, no. 1, pp. 1357–1366.

15. Graaf, M., "Synthesis of the monoacid chloride and the monoalkyl esters of the maleic acid anhydride addition product of l-pimaric acid", *Journal of the American Chemical Society*, 1946, vol. 68, no. 10, pp. 1937–1938.

16. Lewis J. B., Lloyd W. D., Hedrick G. W., "Preparation and some reactions of the vinyl ester of maleopimaric acid", *Journal of Organic Chemistry*, 1960, vol. 25, no. 7, pp. 1206–1208.

17. Huang K., Tang X., Xia J., "Synthesis of trimethyl maleopimaric acid assisted by molecular sieve", *Biomass Chemical Engineering*, 2012, vol. 46, no. 1, pp. 11–14.

18. Hong G., Song Z., Shang S., Xu X., Wang X., "Synthesis, crystal structure and application of maleopimaric acid mono-ethyl ester", *Asian Journal of Chemistry*, 2010, vol. 22, no. 6, pp. 4681–4687.

19. Minmin S., Chengyong H., Zhimeng L., Dongguan Science and Engeeniring, *Glycide maleopimarate and its preparing process*, CN, Pat. № 1297011A PRC, 2001.

20. Qiangqiang Ma, Xiaoqing Liu, Ruoyu Zhang, Jin Zhu, Yanhua Jiang, "Synthesis and properties of full bio-based thermosetting resins from rosin acid and soybean oil: the role of rosin acid derivatives", *Green Chemistry*, 2015, vol. 15, no. 5, pp. 1300–1310.

21. Chernaia N. V., Chubis P. A., "Features of the filler in the sized fiber suspension", *Vestsi Natsyyanal'nai akademii navuk Belarusi. Seryia khimichnykh navuk* [Proceedings of the National Academy of Sciences of Belarus. Chemistry Series], 2008, no. 3, pp. 111–115.

22. Ayman M. Atta, Shymaa M. El-Saeed, Reem K. Farag, "New vinyl ester resins based on rosin for coating application", *Reactive and Functional Polymers*, 2006, vol. 66, no. 12, pp. 1596–1608.

23. Matynia T., Podgorski M., "Synteza i ocena własciwosci multimetakrylanow pochodnych kwasu maleopimarowego", *Przemysl Chemiczny*, 2005, vol. 84, no. 12, pp. 927–932.

24. Jaswal S., Bharti G., "Structure-Property Correlation Study of Bio-Based Multifunctional Vinyl Ester Resin in Presence Of Methacrylated Lignin Model Compounds", *Polymer Science Series B*, 2015, vol. 57, no. 5, pp. 417–433.

25. Popova L. M., Dmitrieva Yu. A., Vershilov S. V., "Preparation of Maleopimar Adducts of Polyfluoroalkyl Esters of Tall Colophony", *Russian Journal of Applied Chemistry*, 2011, vol. 84, no. 5, pp. 895–897.

26. Popova L. M., Vershilov S. V., Dmitrieva Yu. A., "The esterification of maleopimaric adduct with α,α,ω -trihydroper-fluoroalkanols", *Khimiia rastitel nogo syr'ia* [Chemistry of plant raw materials], 2014, no. 1, pp. 73–76.

27. Delevalee F., Deraedt R., Benzoni J., Roussel-UCLAF, Method of inducing immunostimulating activity, US, Pat. № 4880803, 1989.

28. Svikle D. Ia, Kalnin'sh A. Ia, Karklin' R. Ia, Pruse B. A, Prikule A. Ia, Rumba A. A, Rasinia R. A, Shvinska D. F, Baumane G. K, Kul'kevits A. Ia., Institut khimii drevesiny AN LatvSSR, *Soli ili efiry N-zameshchennykh imidov maleopimarovoi kisloty, proiavliaiushchie fungitsidnuiu aktivnost', i sposob ikh poluchenia* [Salts and esters of N-substituted maleopimaric acid imides with fungicide activity and its preparation], Baza patentov SSSR, USSR, Pat. 584722, 1994.

29. Schuller W. H., Lawrence R. W., Culbertson B. M., "Some polyimide-amides from maleopimaric acid", *Journal of Polymer Science*, 1967, vol. 5, no. 8, pp. 2204–2207.

30. Penczek P., "Kwasy N-hydroksyalkilomaleimidopimarowe, N- hydroksyalkilomaleimidopimaric", *Roczniki Chemii*, 1970, vol. 44, no. 10, pp. 1815–1819.

31. Ray S. S., Kundu A. K., Ghosh M., Maiti S., "A new route to synthesize polyamideimide from rosin", *European Polymer Journal*, 1985, vol. 21, no. 2, pp. 131–133.

32. Kazakova O. B., Tret'yakova E. V., Kukovinets O. S., Tolstikov G. A., Nazyrov T. I., Chudov I. V., Ismagilova A. F., "Synthesis and pharmacological activity of amides and the ozonolysis product of maleopimaric acid", *Russian Journal of Bioorganic Chemistry*, 2010, vol. 36, no. 6, pp. 832–840.

33. Jifu Wang, Yung Pin Chen, Kejian Yao, Perry A. Wilbon, Wujie Zhang, Lixia Ren, Juhua Zhou, Mitzi Nagarkatti, Chunpeng Wang, Fuxiang Chu, Xiaoming He, Alan W. Decho, Chuanbing Tang, "Robust antimicrobial compounds and polymers derived from natural resin acids", *Chemical Communications*, 2012, vol. 48, pp. 916–918.

34. Tao Xu, He Liu, Jie Song, Shibin Shang, Zhanqian Song, Kaifei Zou, Chong Yang, "Synthesis and characterization of novel fluorosilicone rubber using imide modified vinyl-containing fluorosilicone resin as cross-linker", *Journal of Polymer Science Part A Polymer Chemistry*, 2015, vol. 53, no. 15, pp. 1769–1776.

35. Tao Xu, He Liu, Jie Song, Shi-Bin Shang, Zhan-Qian Song, Xiu-Jie Chen Chong Yang, "Synthesis and characterization of imide modified poly(dimethylsiloxane) with maleopimaric acid as raw material", *Chinese Chemical Letters*, 2015, vol. 26, no. 5, pp. 572–574.

36. Shengfang Li, Tao Zou, Xianli Liu, Min Tao, "Synthesis and characterization of benzoxazine monomers from rosin and their thermal polymerization", *Designed Monomers and Polymers*, 2013, vol. 17, no. 1, pp. 40–46.

37. Guiyang Yao, Yajun Li, Yongtao Zhu, Yingming Pan, Fuping Huang, Hengshan Wang, Zhixin Liao, "Protonationcontrolled axial chirality in maleopimaric imides", *New Journal of Chemistry*, 2014, vol. 38, no. 2, pp. 693–699.

38. Gui-yang Yao, Man-yi Ye, Ri-zhen Huang, Ya-jun Li, Yong-tao Zhu, Ying-ming Pan, Zhi-Xin Liao, Heng-shan Wangb, "Synthesis and antitumor activity evaluation of maleopimaric acid N-aryl imide atropisomers", *Bioorganic & Medicinal Chemistry Letters*, 2013, vol. 23, no. 24, pp. 6755–6758.

39. Sakhautdinov I. M., Malikova R. N., Zakir'yanova O. V., Abdullin M. F., Yunusov M. S., "Synthesis of a Bromomethylketone Based on N-Maleopimarimide and its Unusual Transformation During Preparation of Sulfonium and Phosphonium Salts", *Chemistry of Natural Compounds*, 2016, vol. 52, no. 1, pp. 78–81.

40. Sakhautdinov I. M., Gumerov A. M., Gibadullina G. G., Zakiryanova O. V., Yunusov M. S., "Synthesis of a 2,3-Dienoate Based on Methyl Maleopimarate", *Chemistry of Natural Compounds*, 2015, vol. 51, no. 2, pp. 383–384.

41. Yu J, Xu N, Liu Z, Wang L., "Novel One-Component Positive-Tone Chemically Amplified I-Line Molecular Glass Photoresists", *ACS Applied Materials & Interfaces*, 2012, vol. 4, no. 5, pp. 2591–2596.

42. Hess S. C., Farah M. I. S., Eguchib S. Y., Imamura P.M., "Synthetic studies with Pinus elliottiis rosin derivatives. Oxidation of maleopimaric anhydride methyl ester and trimethylfumaropimarate", *Journal Of The Brazilian Chemical Society*, 2000, vol. 11, no. 1, pp. 59–63.

43. Nazyrov T. I., Tret'yakova E. V., Kazakova O. B., Spirikhin L. V., Kukovinets O. S., "Regioselective oxidation of the methyl ester of maleopimaric acid by dimethyldioxirane", *Chemistry of Natural Compounds*, 2013, vol. 48, no. 6, pp. 1002–1003.

44. Yao Guiyanga, Wei Jingchena, Dai Weilonga, Yang Daa, Pan Yingminga, Wang Hengshana, "Study on the regioselectivie of Grignard reagent addition reaction of maleopimaric acid trimethyl ester", *Chinese Journal Of Organic Chemistry*, 2013, vol. 33, no. 1, pp. 138–142.

45. Tolstikov A., Karpyshev N., Tolstikova O., Khlebnikova T., Sal'nikov G., Mamatyuk V., Bagryanskaya I., Gatilov Yu., "Derivatives of L-Pimaric Acid in the Synthesis of Chiral Organophosphorus Ligands from Decahydrophenanthrene Series", *Russian Journal of Organic Chemistry*, 2001, vol. 37, no. 8, pp. 1134–1148.

46. Qiang Wu, Gui-yang Yao, Ye Zhang, Heng-shan Wang, Lin Yang, Yong-tao Zhu, Ying-ming Pan, "In situ Synthesis of Rosin Derived Chiral Derivatizing Agents for 31P NMR Assays of Amine and Alcohol Enantiomers", *Chemical Research in Chinese Universities*, 2013, vol. 29, no. 5, pp. 894—899.

47. Hengshan Wang, Xiaoyan Tian, Da Yang, Ying-ming Pan, Qiang Wu, Chunhuan He, "Synthesis and enantiomeric recognition ability of 22-crown-6 ethers derived from rosin acid and BINOL", *Tetrahedron Asymmetry*, 2011, vol. 22, no. 4, pp. 381–386.

48. Liu Lu-zhi, He Chun-huan, Yang Lin, Huang Yan, Wu Qiang, Duan Wen-gui, Wang Heng-shan, Pan Ying-ming, "Novel C1-symmetric chiral crown ethers bearing rosin acids groups: synthesis and enantiomeric recognition for ammonium salts", *Tetrahedron*, 2014, vol. 70, no. 50, pp. 9545–9553.

49. Bei M. P., Azarko V. A., Yuvchenko A. P., "Synthesis, Film-Forming, and Light-Sensitivity Properties of Allyl and Propargyl Maleopimarates and Cytraconopimarates", *Russian Journal of General Chemistry*, 2010, vol. 80, no. 5, pp. 940–943.

50. Bei M. P., Yuvchenko A. P., "The synthesis of new maleopimaric acid amides and imides", *Vestsi Natsyyanal'nai akademii navuk Belarusi. Seryia khimichnykh navuk* [Proceedings of the National Academy of Sciences of Belarus. Chemistry Series], 2010, no. 1, pp. 74–78.

51. Bei M. P., Yuvchenko A. P., Puchkova N. V., "Efficient Synthesis of Maleopimaric Acid N-Arylimides", *Russian Journal of General Chemistry*, 2015, vol. 85, no. 5, pp. 1034–1039.

52. Bei M. P., Yuvchenko A. P., "The new secondary terpenoid products based on renewable wood chemistry raw materials", *XIX Mendeleevskii s"ezd po obshchei i prikladnoi chimii, Volgograd, 25-30 sentiabria, 2011. Tezisy dokladov.* [XIX Mendeleev congress on general and applied chemistry, Volgograd, September 25-30, 2011. Book of abstracts], IUNL VolgGTU, Volgograd, RU, 2011, vol. 4, p. 185.

53. Bei M. P., Yuvchenko A. P., "Synthesis and Properties of Maleopimaric N-(n-Alkyl)imides", *Russian Journal of General Chemistry*, 2010, vol. 80, no. 2, pp. 253–257.

54. Bei M. P., Yuvchenko A. P., "The Synthesis of trans-1,2-Dicarboxylic Acids from Maleopimaric Acid Monoamides", *Vestsi Natsyyanal'nai akademii navuk Belarusi. Seryia khimichnykh navuk* [Proceedings of the National Academy of Sciences of Belarus. Chemistry Series], 2010, no. 3, pp. 84–87.

55. Bei M. P., Yuvchenko A. P., Gosudarstvennoe nauchnoe uchrezhdenie "Institut khimii novykh materialov Natsional'noi akademii nauk Belarusi", *Sposob polucheniia tsitrakonopimarovoi kisloty* [The Method for Preparation of Citraconopimaric Acid], Natsional'nyi tsentr intellektual'noi sobstvennosti, BY, Pat. 13646, 2010.

56. Bei M. P., Baranovsky A. V., Yuvchenko A. P., "Structure of a rosin-itaconic acid adduct from 2D NMR spectroscopy data", *Journal of Applied Spectroscopy*, 2009, vol. 76, no. 4, pp. 603–606.

57. Koroleva E. V., Gusak K. N., Ignatovich Zh. V., Ermolinskaya A. L., Bei M. P., Yuvchenko, "Synthesis of maleopimaric and citraconopimaric acids N-[3-(pyrimidin-2-yl)aryl]amides", *Russian Journal of Organic Chemistry*, 2012, vol. 48, no. 8, pp. 1121–1125.

58. Yuvchenko A. P., Bei M. P., "The Synthesis, Properties and Use of New Secondary Terpenoid Products from Rosin", *Vestsi Natsyyanal'nai akademii navuk Belarusi. Seryia khimichnykh navuk* [Proceedings of the National Academy of Sciences of Belarus. Chemistry Series], 2013, no 4, pp. 68–74.

59. Bei M. P., Yuvchenko A. P., Baranovsky A. V., "Formation of N-Alkylimides in Reaction of Maleopimaric and Citraconopimaric Acids with Secondary Amines", *Vestsi Natsyyanal'nai akademii navuk Belarusi. Seryia khimichnykh navuk* [Proceedings of the National Academy of Sciences of Belarus. Chemistry Series], 2013, no 4, pp. 104–108.

60. Pesetskii S. S., Krivoguz Iu. M., Yuvchenko A. P., Bei M. P., Gosudarstvennoe nauchnoe uchrezhdenie "Institut mekhaniki metallopolimernykh sistem imeni V. A. Belogo Natsional'noi akademii nauk Belarusi", Gosudarstvennoe nauchnoe uchrezhdenie "Institut khimii novykh materialov Natsional'noi akademii nauk Belarusi", *Sposob polucheniia adgesiva* [The Method for Preparation of Adhesive], Natsional'nyi tsentr intellektual'noi sobstvennosti, BY, Pat. 14660, 2011.

61. Petrushenia A. F., Reviako M. M., Iatsenko V. V., Bei M. P., Iuvchenko, A. P., "Analysis of the Influence of Secondary Terpenoid Products on the Adhesion of the PE", *Vestsi Natsyyanal'nai akademii navuk Belarusi. Seryia fizika-tekhnichnykh navuk* [Proceedings of the National Academy of Sciences of Belarus. Physico-technical series], 2012, no. 3, pp. 21–24.

62. Muravskii A. A., e. a., *Chiral'naia zhidkokristallicheskaia kompozitsiia* [Chiral LC-composition], BY, Pat. a20131038, 2013.

63. Yuvchenko A. P., Goncharova I. A., Bei M. P., Lugovneva A. P., Agabekov V. E., Kulevskaia I. V., Gosudarstvennoe nauchnoe uchrezhdenie "Institut khimii novykh materialov Natsional'noi akademii nauk Belarusi", Gosudarstvennoe nauchnoe uchrezhdenie "Institut mikrobiologii Natsional'noi akademii nauk Belarusi", *Antisepticheskaia kompozitsiia dlia zashchity celliulozosoderzhashchikh materialov* [Antiseptic composition for protection of cellulose materials], Natsional'nyi tsentr intellektual'noi sobstvennosti, BY, Pat. 16543, 2012.

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